ALKALOIDS FROM DENDROBATID POISON FROGS: FURTHER PUMILIOTOXINS AND ALLOPUMILIOTOXINS AND A REASSIGNMENT OF THE KETO FUNCTION IN PUMILIOTOXIN 307F

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Abstract - Skin extracts from the Panamanian poison-frog Dendrobates pumilio have afforded further trace alkaloids of the pumiliotoxin-A class (6-alkylidene-8hydroxy-8-methylindolizidines) and an allopumiliotoxin subclass (7-hydroxy-PTX-As). Structures for pumiliotoxins 209F and 225F and allopumiliotoxin 225E, all with four carbon 6-alkylidene side chains, for allopumiliotoxins 309D, 325A' and 325A'', all with ten carbon 6-alkylidene side chains, and a reassignment of the position of the keto function in the ten carbon 6-alkylidene side chain of pumiliotoxin 307F are presented. Carbon-13 magnetic resonance assignments for these and the 15-0-methyl ether artefacts of pumiliotoxin 307A and allopumiliotoxin 323B, are tabulated.

Dendrobatid alkaloids of the pumiliotoxin-A class exhibit remarkable myotonic and cardiotonic activity (1,2) with activity being optimal in pumiliotoxin B (3). Alkaloids of the pumiliotoxin-A class have been divided into three subclasses, the pumiliotoxins, the allopumiliotoxins and the homopumiliotoxins. The pumiliotoxins exhibit a characteristic mass spectral pattern, showing a loss of OH ($M^{+}-17$), and major ions at m/z194 (C12H20NO⁺), 166 (C10H16NO⁺) and 70 (C4H8N⁺). The structural elucidation of pumiliotoxin A and B remained a challenge for nearly a decade after initial isolation and characterization in 1967 (4). The structure of a simpler alkaloid, pumiliotoxin 251D was defined by x-ray crystallographic analysis of a crystal of 251D.HCl and the structures of pumiliotoxin A (307A) and pumiliotoxin B (323A) and of other pumiliotoxins and allopumiliotoxins were deduced from analysis of their mass and NMR spectra, as compared to that of pumiliotoxin 251D (5). The allopumiliotoxins were hydroxy congeners of the pumiliotoxins, containing a 7-hydroxy group in the indolızidine moiety. The allopumiliotoxins exhibit a characteristic mass spectral pattern, showing a loss of OH $(M^{+}-17)$, and major ions at m/z 210 $(C_{12}H_{20}NO_{2}^{+})$, 182 $(C_{10}H_{16}NO_{2}^{+})$ and 70 $(C_{4}H_{8}N^{+})$. Some seventeen pumiliotoxins and fifteen allopumiliotoxins have been identified from extracts of skins of dendrobatid frogs. The structures in many cases, were based only on mass spectral analysis of the alkaloid before and after deuterium exchange with ND3, and of a

perhydro-derivative of the alkaloid produced by catalytic hydrogenation (6). Only one member of a homopumiliotoxin subclass, 223G, has been isolated and characterized (7). The homopumiliotoxins differ from the pumiliotoxins in being quinolizidines rather than indolizidines. The mass spectrum of a homopumiliotoxin has major ions at m/z 180 (C₁₁H₁₈NO⁺), and 84 (C₅H₁₀N⁺) originating from the 7-alkylidene-9-hydroxy-9-methylquinolizidine moiety. Mass, NMR and IR spectral analyses now have defined the structures for additional trace pumiliotoxin-A class alkaloids from extracts of skin of the Panamanian poison frog *Dendrobates pumilio* (For prior spectral analyses of pumiliotoxin-A class alkaloids see ref. 5,7,8). Structures of dendrobatid alkaloids discussed in the present paper are presented in Fig. 1.

<u>Pumiliotoxins</u>

The alkaloids 209F and 225F exhibit mass spectra characteristic of pumiliotoxins with major fragments at m/z 166 and 70. The NMR spectra also were characteristic of pumiliotoxins and led to definition of the structures of 209F and 225F. The ¹³C-NMR assignments for pumiliotoxins 209F and 225F are presented in Table 1, along with those of pumiliotoxin 251D, which has a seven carbon alkylidene side chain and whose structure has been defined by x-ray crystallographic analysis (5). The ¹H-NMR spectrum of 209F had well separated distinct doublets at δ 0.99 (J = 6.8 Hz) and 0.91 (J = 6.6 Hz) due to the diastereotopic methyls of the gem dimethyl group. The two methyl groups showed NOEs of 2% and 3% with the 10-hydrogen. The ¹³C- and the ¹H-NMR spectra of pumiliotoxin 225F clearly define the structure as that of a 12-hydroxy-209F.

The GC-FTIR spectra of alkaloids of the pumiliotoxin-A class are proving extremely useful for characterization of such dendrobatid alkaloids. FTIR spectra of pumiliotoxin A (307A), pumiliotoxin B (323A) and allopumiliotoxin B (323B') are presented in Figs. 2 and 3. A GC-FTIR spectrum of 209F (Fig. 2) revealed a sharp peak at 3544 cm^{-1} typical of the hydroxyl OH stretching frequency of a hydrogen-bonded tertiary alcohol common to the pumiliotoxin-A class. A characteristic Bohlmann band pattern at 2798-2700 cm^{-1} and a sharp absorption at 963 cm⁻¹ (C(6)-C(10) trisubstituted double bond) also are typical of alkaloids of the pumiliotoxin-A class. However, the 990 cm^{-1} absorption of PTX-A, which we assign to a carbon-hydrogen out-of-plane vibration of the C(13)-C(14) double bond, is absent in the IR spectrum of 209F. The IR spectrum of 225F showed the same hydroxyl stretching absorption as 209F at 3543 cm⁻¹. In addition, there were two weaker hydroxyl stretching absorptions at approximately 3650 and 3600 cm⁻¹. The latter two absorptions dísappear on formation of a mono-O-acetate (Py:Ac20 (1:1); 2 h), whose IR spectrum showed 3543, 1761, 1229 cm⁻¹ among other absorptions (Fig. 2). The spectrum of 225F also showed Bohlmann bands at 2799-2700 cm⁻¹, a C-O stretching absorption at 1035 cm⁻¹ and the trisubstituted double bond C-H deformation absorption at 965 cm⁻¹. The data indicate that the second hydroxyl group at 225F is acetylatable (i.e. non-tertiary) and that it is partially hydrogen-bonded (see Fig. 2).

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Fig. 1. Proposed structures of pumiliotoxins 209F and 225F, allopumiliotoxins 225E, 309D, 325A' and 325A" and a revised structure for pumiliotoxin 307F. A tentative structure for pumiliotoxin 307B, the structures of pumiliotoxins 251D, 307A', 307A", 323A and erythro-PTX-B, allopumiliotoxins 323B' and 323B", and the 15-0-methyl ethers of 307A (alkaloid "321", ref. 7) and 323B also are shown. The numbering of the simpler alkaloids, 209F, 225F, etc., retains the numbering of the C-18 methyl group for NMR comparison purposes only. TABLE 1 Carbon 13-magnetic resonance assignments for pumiliotoxins 209F, 225F and 251D. Assignments for 251D from ref. 8. Assignments for 209F and 225F were based on hetero- and homo-nuclear shift correlation spectroscopies (Solvent CDCl₃, 40°C, 100 MH₂). Values marked with asterisks may be transposed. The values for 209F agree with those of synthetic material (L.E. Overman, personal communication).

Carbon	209F	225 F	251D
1	23.3.t	23.1.t	23.4.t
2	21.2,t	21.1,t	21.2,t
3	54.6,t	54.3.t	54.7.t
5	53.0,t	53.3,t	53.3,t
6	129.4,s	133.9,s	130.0,s
7	48.8,t	49.0,t	49.0,t
8	68.4,s	68.4,s	68.4,s
8a	71.8,d	72.0,đ	71.8,d
9	24.3,q	24.5,q	24.3,q
10	135.7,d	130.7,a	134.7,d
11	26.8,d	35.1,d	32.1,d
12	23.4,q*	17.1,g	37.6,t
13		• •	29.8.t
14			22.9.t
15			14.2,q
18	23.5,q*	67.9,t	21.8,q

<u>Allopumiliotoxins</u>

Allopumiliotoxin 225E was not isolated in quantities sufficient for ¹³C-NMR assignments, but the mass spectrum indicates it has a structure compatible with 7-hydroxy-209F. The ¹H-NMR spectrum for 225E confirms this assignment and demonstrates the usual 7,8-trans-diaxial hydroxyl groups of the allopumiliotoxins (data not shown). A GC-FTIR spectrum of 225E was not obtained.

Three new allopumiliotoxins with ten carbon alkylidene side chains were isolated in small quantities from the *Dendrobates pumilio* extracts. These were allopumiliotoxins 309D, 325A' and 325A" (Fig. 1). The latter two alkaloids appear, based on NMR spectral analysis, to be 13,14-dihydro analogs of allopumiliotoxins 323B' and 323B". The stereochemistry at C-14 and C-15 is unknown for 325A' and 325A". ¹³C-NMR assignments are presented in Table 2, along with previous assignments for 323B' and 323B". A tentative structure for 309D based on NMR spectral analysis and comparison of its 6-alkylidene moiety to that of the 325A isomers, is presented in Fig. 1 (see Table 2 for ¹³C-NMR assignments).

An FTIR spectrum of 325A' shows two equivalent, sharp hydroxyl OH stretching frequencies at 3649 and 3522 cm⁻¹, the latter typical of the tertiary hydrogen-bonded hydroxyl in the allopumiliotoxins (Fig. 3). A sharp Bohlmann band at 2802 and a pair of absorptions at 1013 and 977 cm⁻¹ are seen. The absorption at 977 cm⁻¹ is consonant with the presence of the exocyclic C(6)-C(10) double bond. Unfortunately the C-O stretching frequency apparently overlaps the δ_{C-H} of the C(13)-C(14) double bond so the absence of the C(13)-C(14) double bond is not clear from the FTIR spectrum. The intensity of this •



Fig. 2 FTIR spectra of pumiliotoxins 209F, 225F, 307B, pumiliotoxin A (307A') and the 12-0-acetate of 225F.

absorption (1011 cm⁻¹), however, does appear to be reduced in 325A' relative to 323B' (see Fig. 3). The most unambiguous evidence for the saturation of this bond rather than the C(6)-C(10) double bond in 325A' is the mass spectrum, which shows an m/z 182 base peak. FTIR spectra of 325A" and 309D were not obtained.



Fig. 3 FTIR spectra of pumiliotoxin 307F and a minor isomer of 307F, pumiliotoxin B (323A), erythro-pumiliotoxin B, and allopumiliotoxins 323B' and 325A'.

Carbon	309D	323B'	323B"	3258'	3258"	15-0-methyl-323B
1	22.7,t	22.7,t	22.7,t	22.7,t	22.7,t	22.9,t
2	21.3,t	21.3,t	21.2,t	21.3,t	21.3,t	21.5,t
3	54.3,t	54.3,t	54.3,t	54.3,t	54.3,t	54.4,t
5	49.0,t	49.3,t	49.0,t	49.1,t	49.2,t	49.1,t
6	133.8,s	133.8,s	133.7,s	133.8,s	133.8,s	133.3,s
7	81.0,d	80.9,d	80.8,d	80.9,d	80.9,đ	80.8,d
8	70.3,s	70.4.s	70.4,s	70.3,s	70.3,s	70.4,s
8a	65.3,d	65.2,d	65.3,d	65.3,d	65.3,d	65.4,d
9	20.6.g	20.6.g	20.6,q	20.6,q	20.6,q	21.0,q
10	138.8,d	137.3.d	137.8,d	138.4,d	138.3,d	137.7,d
11	32.3,d	32.6,d	32.4,d	31.8,d	32.5,d	32.7,d
12	29.7,t	35.4,t	35.1,t	30.5,t	29.7,t	35.4,t
13	34.8.t	125.8.d	124.2.d	34.1,t	35.2,t	126.3,d
14	32.5,d	137.6,s	138.0,s	37.4,d	37.6,d	137.6,s
15	39.4.t	80.2.d	79.0,d	77.2,d	76.7,d	89.2,d
16	20.2.t	27.8.t	27.8.t	26.9.t	26.7.t	26.5,t
17	14.4,q	10.2.g	10.2.σ	10.7,q	10.7,q	10.7,q
18	21.3,q	20.9,q	20.9,q	21.5,q	21.3,q	21.1,q
19	19.7.a	10.9.a	12.2.g	15.3,q	15.3,q	11.1.g

TABLE 2 Carbon 13-magnetic resonance assignments for allopumiliotoxins 309D, 323B', 323B", 325A', 325A" and 15-0-methyl-323B. Assignments for 323B' and 323B" are from ref. 8. (Solvent CDCl₃, 40°C, 100 MH₂)

-OMe 56.0,q

The ¹³C-NMR assignments for an O-methyl derivative of allopumiliotoxin 323B also are presented in Table 2. Such O-methyl derivatives form easily from certain of the pumiliotoxin-A class alkaloids that contain an allylic 15-hydroxy group. For example, pumiliotoxin A (307A) yields an O-methyl ether, previously thought to occur naturally as pumiliotoxin 321 (7), but now recognized as an artefact. Such O-methyl derivatives <u>do not</u> form from alkaloids that contain the 15,16-dihydroxy moiety, as in pumiliotoxin B (323A) and allopumiliotoxin 339A (unpublished results). Epimerization at the 15-position occurs readily under acid conditions for 307A and for 323B and has led to isolation of both 15epimers (307A', 307A'', 323B', 323B'') for these alkaloids (8). Epimerization at the 15position of pumiliotoxin B (323A) has not yet been attained, in spite of strenuous conditions, presumably due to a stabilizing effect of the 16-hydroxyl group (see ref. 9).

Pumiliotoxin A (307A) presumably occurs naturally as 307A' with the same configuration of the 15-hydroxyl group as in pumiliotoxin B (323A). Epimerization at C(15) during isolation yields varying amounts of 307A", which can be separated from 307A' only by HPLC and which has a mass spectrum identical to 307A'. Since much of our preliminary characterization of frog skin extracts is done with capillary GC-MS where 307A' and 307A" co-chromatograph, we reserve the 307A identification to refer to material that may or may not be a mixture of 15-hydroxy diastereomers. The same situation holds for allopumiliotoxin-B (323B) and the methyl ether artefact, 15-0-methyl-323B. In addition to epimerization at C(15) during isolation, it appears likely that pumiliotoxin A

Carbon	307A'	307A"	307 F	
1	23.3,t	23.3,t	23.1,t	
2	21.2.t	21.2,t	21.0,t	
3	54.6.t	54.6.t	54.4,t	
5	53.3.t	53.3.t	53.0,t	
6	130.3,s	130.4,s	130.6,s	
7	48.9.t	48.9,t	48.7,t	
8	68.4,s	68.4,s	68.3,s	
8a	71.7,d	71.7,đ	71.6,d	
9	24.3,q	24.3,q	24.2,q	
10	133.8,d	133.8,d	132.6,d	
11	32.6,d	32.6,d	28.0,d	
12	35.6,t	35.6,t	48.3,t	
13	125.0,d	125.0,d	213.6,s	
14	137.7,s	137.7,s	46.6,d	
15	79.6,d	79.5,d	34.8,t	
16	27.8,t	27.8,t	20.3,t	
17	10.2.q	10.2,q	14.1,q	
18	21.3,q	21.4,q	21.5,q	
19	11.4,q	11.4,q	15.9,q	

TABLE 3 Carbon 13-magnetic resonance assignments for pumiliotoxins 307A', 307A" and 307F. Assignments for 307A' and 307A" are from ref. 5. (Solvent CDCl₃, 40°C, 100 MH₂)

also undergoes allylic rearrangement to yield another isomer. Such an isomer has been isolated in trace amounts by HPLC from the *Dendrobates pumilio* extract. The mass spectrum and the slightly shorter retention time on packed or capillary (OV-1) columns than either 307A' or 307A'' indicate that it is 307B, an alkaloid often reported as a trace constituent in dendrobatid extracts that contain major amounts of pumiliotoxin A (see ref. 6). The mass spectrum of 307B had a ratio of m/z 193 to m/z 194 of two to one (see ref. 6) similar to that observed with the present trace alkaloid. In contrast, the m/z 194 peak for 307A was much larger than the m/z 193 peak. The FTIR of this trace alkaloid is very similar to that of 307A' (see Fig. 2). At present we consider this trace alkaloid, 307B, to correspond to an acid-catalyzed allylic rearrangement product of 307A and perhaps, like the 15-0-methyl ethers to be an artefact of the isolation procedure.

Alkaloids 307B and 307F have similar mass spectra and cochromatograph on the packed gas chromatography columns used in earlier analyses. Thus, it is likely that the designation 307B applied to some extracts may have been used to characterize an alkaloid peak comprised of a mixture of 307B and 307F or even of 307F rather than 307B. Indeed, "307B" in some extracts showed only one exchangeable hydrogen rather than two (6). FTIR spectra clearly distinguish 307B and 307F and moderately polar capillary columns allow their separation.

A Revised Structure for Pumiliotoxin 307F

The structure proposed (7) for **307F** has proven on further nuclear magnetic resonance analysis to be incorrect. The keto group of this pumiliotoxin is at the 13-position,

rather than at the 15-position as initially proposed. The reassignment to the 13-position is based on HOHAHA experiments, which exhibited mutual magnetization transfer relay between both the C-17 and C-19 methyl groups. The revised ¹³C-NMR assignments for pumiliotoxin 307F are presented in Table 3.

The GC-FTIR chromatogram of 307F shows two partially resolved peaks with identical IR spectra, both exhibiting a strong OH stretching absorption at 3542 cm⁻¹ and a carbonyl absorption at 1720 cm⁻¹ (see Fig. 3). The Bohlmann band pattern at 2799-2700 cm⁻¹ and olefin absorption at 963 cm⁻¹ are typical of the pumiliotoxin-A class of dendrobatid alkaloids. The two partially resolved GC peaks (~1:1) are assumed to comprise a mixture of the 14-methyl epimers of 307F. A minor peak following the main pair also has a very similar IR spectrum and may be the 15-ketone originally proposed as the structure for 307F, as shown in Fig. 3.

A sample of 307F exposed to excess NaOCD₃ in CD₃OD (1 h, RT) gave an approximately 1:1 mixture of bis and tris deuterium-exchanged material whose E.I. M.S. cleavage pattern supports the location of the ketone at C-13. In particular, fragments at m/z 237/238 (undeuterated 307F had m/z 236) and m/z 208/209/210 (m/z 208 in undeuterated 307F) indicate ketonic cleavage at the C(13)-C(14) and C(12)-C(13) bonds, respectively.

EXPERIMENTAL

High-resolution mass spectral data were obtained on a JEOL D-300 mass spectrometer (electron impact, 70 eV). Gas chromatographic mass spectral analyses were with a 2% OV-225 Chromosorb G AW-DMCS column programmed from 100° to 250° at 10°/min (electron impact, 30 eV). Nuclear magnetic resonance spectra were obtained on JEOL FX-100 and GX-400 spectrometers. Phase-sensitive 2D-NMR, homo- and hetero-nuclear shift correlation spectroscopies were routinely used for structure analysis. HOHAHA and NOE effects were observed through 2D-spectra and 1D-differential spectra. A Hewlett-Packard model 5965A FTIR instrument with a narrow band (4000-750 cm⁻¹) detector and a 59970 IRD ChemStation was used to record FTIR spectra of GC peaks. A Hewlett-Packard model 5890 gas chromatograph fitted with an HP-5 (bonded 5% diphenylsiloxane : 95% dimethylsiloxane) fused silica capillary column (30 m x 0.32 mm) programmed from 100° to 280° at 10°/min was used to generate the total response chromatogram.

Isolation of Alkaloids

A methanolic extract from 1200 skins of *Dendrobates pumilio* from Isla Bastimentos, Bocas, Panama (1987) was partitioned between aqueous methanol-chloroform. Alkaloids were then extracted from the chloroform layer with 0.05 N HCl. After adjusting to pH > 10 with ammonia, the aqueous layer was extracted with hexane and then chloroform. The hexane and chloroform layers were evaporated in vacuo to afford 0.13 g and 0.18 g, respectively. The hexane extract residue was treated with hexane containing 0.3% triethylamine. The soluble portion was chromatographed on a DIOL column (Merck, prepacked Lobar column, size B) with hexane-0.3% triethylamine to afford five main fractions (H-1 to H-5) as monitored by refractive index. GC-MS analysis revealed the following main alkaloids from each fraction: Fraction H-1, 20 mg, cis-195A and 205B; H-2, 20 mg, cis-195A; H-3, 8 mg, 209F and 205A; H-4, 5 mg, mainly 235B; H-5, 5 mg, mainly 321. Fraction H-3 was further purified by HPLC on an ODS column with acetonitrile-water-triethylamine (60:40:1) to provide 2 mg 209F and 3 mg 205A. The chloroform extract residue was chromatographed on a reversed phase silica gel column (Merck, prepacked Lobar column, RP-8, size B) with acetonitrile-water-triethylamine (40:60:1) to afford five main fractions (C-1 to C-5). GC-MS analysis revealed the following: Fraction C-1, 14 mg, mainly the O-methyl ether of 323B; C-2, 12 mg, 225E and 225F; C-3 110 mg, 323A; C-4, 16 mg, 323B; C-5, 23 mg, cis-211A,

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323B and 325A. Rechromatography of C-2 with hexane-chloroform-isopropanol-triethylamine (60:40:1:1) afforded almost pure samples of 2 mg 225E and 3 mg 225F. Rechromatography of C-5 with hexane-chloroform-triethylamine (50:50:1) yielded almost pure samples of cis-211A (3 mg), 323B (4 mg) and 325A (3 mg). The latter material on rechromatography provided 325A' and 325A".

Extracts from Panamanian poison frog *Dendrobates pumilio* (Isla Bastimentos, Bocas, Panama) contain more than forty alkaloids (ref. 6,7 and unpublished data). These include pumiliotoxin A (307A) and pumiliotoxin B (323A) and other pumiliotoxins, allopumiliotoxins, decahydroquinolines, 5-substituted-8-methylindolizidines and alkaloids of unknown classes. Properties of 209F, 225E, 225F, 309D, 325A', 325A'' and the 15-0methyl ether of 323B from *Dendrobates pumilio* are presented below. The empirical formulas as determined by high resolution mass spectrometry are followed by the electron impact mass spectra with intensities relative to a base peak set equal to 100 in parentheses and the optical rotations if known.

Pumiliotoxin 209F, $C_{13}H_{23}NO$, m/z 209 (22), 166 (70), 70 (100), $[\alpha]_D$ -11.6° (c 0.10 CHCl₃).

Allopumiliotoxin 225E, C13H23NO2, m/z 225 (40), 208 (86), 182 (41), 138 (41), 114 (34), 112 (60), 70 (100).

Pumiliotoxin 225F, C13H₂₃NO₂, m/z 225 (21), 194 (28), 166 (79), 112 (19), 84 (26), 70 (100), $[\alpha]_D$ -87.4° (c 0.23, CHCl₃).

Allopumiliotoxin 309D, $C_{19}H_{35}NO_2$, m/z 309 (13), 292 (34), 182 (37), 123 (22), 114 (26), 112 (35), 70 (100). This alkaloid was isolated from a 1983 collection of *D. pumilio* at the same site (7), but was lost during purification after initial NMR analyses. It was not detected in the 1987 extract.

Allopumiliotoxin 325A', $C_{19}H_{35}NO_3$, m/z 325 (12), 308 (22), 182 (100), 114 (25), 112 (21), 70 (73). An isomer allopumiliotoxin 325A" ($C_{19}H_{35}NO_3$), was also isolated. The mass spectrum was virtually identical to that of 325A'.

15-O-methyl ether of allopumiliotoxin **323B**, $C_{20}H_{35}NO_3$, m/z 337 (16), 322 (11), 306 (69), 288 (23), 262 (10), 222 (17), 209 (59), 192 (29), 182 (100), 114 (44), 70 (73), $[\alpha]_{\rm D}$, + 24.0° (c 0.2, CHCl₃).

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